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AP Chemistry Lesson for Don Bosco Technical Institute. This lesson covers the process of dissolution, conductivity of solutions, pH, and net ionic equations.

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aqueous solutions ap chemistry 20 general properties a solution in which water is the dissolving medium is called an aqueous solution In this unit, we examine the chemical reactions taking place in this aqueous solutions A solution is a

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AP Chemistry Name ____ Period ____ Date __/__/__ 4 Reactions in Aqueous Solution M O L A R I T Y Notes: Concentration can be measured in terms of molarity (M). Molarity = solution of Liters solute of moles also moles of solute = (Molarity) (Liters of solution) Dilution problems can be solved with the formula: V·M = V·M Molarity can be used as a conversion factor to convert moles to Liters of ...

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• A solution is a homogeneous mixture of two or more substances. • A solution is made when one substance (the solute) is dissolved in another (the solvent). • The solute is the substance that is present in smallest amount. • Solutions in which water is the solvent are called aqueous solutions.

[Common Student Misconceptions](#)

aqueous solution - solutions in which water is the dissolving medium water able to dissolve wide assortment of substances; most chemical reactions require water to take place; solution - homogeneous mixture of 2+ substances; solvent - substance in the solution in greater quantity ; solute - other substances besides the solvent in a solution

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(a) A solution containing 3.23 grams of an unknown compound dissolved in 100.0 grams of water freezes at -0.97°C. The solution does not conduct electricity.

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AP Chemistry - Molecular, Ionic and Net Ionic Equations. AP Chemistry: Worksheet--Molecular, Ionic and Net Ionic Equations. 1. Strontium compounds are often used in flares because their flame color is bright red. One industrial process to produce low-solubility strontium compounds (that are less affected by getting wet) involves the reaction of aqueous solutions of strontium nitrate and sodium carbonate.

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[Ap Chemistry Practice Test: Chapter 4 SHORT ANSWER. Write](#)

An aqueous solution is any solution in which water (H 2 O) is the solvent.. In a chemical equation, the symbol (aq) follows a species name to indicate that it is in aqueous solution.For example, dissolving salt in water has the chemical reaction:

[Aqueous Solution Definition in Chemistry](#)

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