

Chapter 7test Chemical Formulas And Compounds

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Chapter 7: Chemical Formulas and Chemical Compounds You'll ...

CHAPTER 7 REVIEW Chemical Formulas and Chemical Compounds SECTION 2 SHORT ANSWER Answer the following questions in the space provided. 1. Assign the oxidation number to the specified element in each of the following examples: 4 a. S in H_2SO_3 6 b. S in MgSO_4 2 c. S in K_2S 1 d. Cu in Cu_2S 6 e. Cr in Na_2CrO_4 5 f. N in HNO_3 4 g. C in (HCO_3) 3 h. N in (NH_4) 2. SCl_2 2 a.

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Chapter 7 Chemical Formulas And Compounds Test B Answers

WRITING CHEMICAL FORMULAS PRACTICE □ Write the formula of the binary ionic compound that will result between the following elements: □ 1. zinc and sulfur □ 2. aluminum and phosphorus □ 3. strontium and chlorine □ 4. barium and oxygen □ 5. sodium and nitrogen □ 6. lithium and iodine

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Chapter 7 Notes □ Chemistry; Chemical Formulas and Chemical Compounds. Molecules and Molecular Compounds. □ Amoleculeconsists of two or more atoms bounded together. Molecules and Chemical Formulas. □ Each molecule has a chemical formula. □ The chemical formula indicates: 1. Which atoms are found in the molecule, and. 2.

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Chemistry Chapter 7 Test Chemical Formulas And Compounds

Chapter 7: Chemical Formulas and Chemical Compounds Section 1: Chemical Names and Formulas

Overview We will explain the significance of a chemical formula. We will again look at how to determine the formula unit of an ionic compound. We will name an ionic compound based on its formula unit. We will use prefixes to name binary molecular ...

Chapter 7: Chemical Formulas and Chemical compounds

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with which they undergo certain chemical reactions. 20. A(n) uses symbols and formulas to represent the names and relative amounts of reactants and products in a chemical reaction. 21. The law of must be satisfied for a chemical equation to be balanced. 22. A(n) product is sometimes indicated by an arrow pointing upward. 23.

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